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Research Article

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Ecotoxicological effects of iron on the activities of antioxidant enzymes in Safflower (*Carthamus tinctorius* L.) seedlings

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ABSTRACT

Iron is an essential metal at micro-concentration for the growth of plants and their metabolic activities. Excess Fe^{2+} concentration generate oxidative stress within sub cellular compartments of the plant cell which are derived from the metabolism of oxygen. To combat the oxidative damage plants have the antioxidant defense system to remove, neutralize and scavenge the Reactive Oxygen Species (ROS) which is produced during heavy metal stress. The paper examines the effect of excess of $FeSO_4$ on the activities of some antioxidative enzymes, catalase and peroxidase in roots and shoots of Safflower seedlings. Fe^{2+} solution treated Safflower seeds with different concentrations were sown in plastic pots and incubated at greenhouse conditions. Fresh roots and shoots were collected at different intervals to evaluate the enzyme activity. Protein estimation and enzyme assay were carried out using standard methods. The activity of peroxidase (POX) and catalase (CAT) of Carthamus tinctorius treated with Fe^{2+} mainly displayed biphasic responses with increased metal concentration. In the present investigation, the excess of Fe^{2+} $(100-200 \text{ mg},l^{1})$ induced toxicity in root and shoot of C. tinctorius. The results indicated an enhancement in the activity of peroxidase in shoots, suggesting that this enzyme serves as an intrinsic defense tool to resist Fe-induced oxidative damage in safflower plants, while catalase did not appear to be an efficient scavenger of hydrogen peroxide (H_2O_2) than POX. However, results indicated that the exposure of C.tinctorius to iron provoke pronounced responses of antioxidative systems which protects the plants to some extent against oxidative damage.

Key words: Antioxidant, Catalase, Peroxidase, Toxicity, Biphasic.

INTRODUCTION

It is well documented that iron is an essential metal at micro concentrations; however, excess concentrations are believed to generate oxidative stress as understood by an increase in the steady state concentration of reactive oxygen species (ROS) within subcellular compartments of the plant cell which are derived from the metabolism of oxygen. Reactive oxygen species which include superoxide anion radical (O_2^{-}), hydrogen peroxide (H_2O_2), hydroxyl radical (OH⁻) and singlet oxygen (1O_2), are amongst the most reactive compounds known to be produced during heavy metal stress¹⁰.

To combat the oxidative damage plants have the antioxidant defense system comprising of enzymes - catalase (EC 1.11.1.6), peroxidases (EC 1.11.1.7), superoxide dismutases (SOD, EC 1.15.1.1) and the nonenzymic constituents alpha-tocopherol, ascorbate and reduced glutathione which remove, neutralize and scavenge the ROS^{21} . Catalases are involved in scavenging H_2O_2 generated during the photorespiration and beta-oxidation of fatty acids¹⁸. Peroxidases are heme containing proteins that utilize H_2O_2 in the oxidation of various organic and inorganic substrates¹. Peroxidases utilizing guaiacol as electron donor in vitro are guaiacol peroxidases and participate in developmental processes, lignification, ethylene biosynthesis, defense, wound healing, etc².

Devi Prasad, A. G. et al Int. J. Pure App. Biosci. **2** (5): 118-123 (2014) ISSN: 2320 – 7051

The other groups of peroxidases scavenge H_2O_2 in cell and utilize glutathione, Cytochrome, pyridine nucleotide and ascorbate as electron donors *in vitro*². Guaiacol peroxidases are glycoproteins, located in cytosol, vacuole, cell wall and in extracellular space, while the other group is non glycosylated and localized in chloroplasts and cytosol².

Safflower (*Carthamus tinctorius* L.) is a tap-rooted annual crop which can tolerate environmental stresses including salinity and water stress¹³. After the commercialization of less expensive synthetic dyes (aniline), safflower has been grown predominantly for its seed oil¹². Since safflower is more drought and salt tolerant than some other oilseed plants such as sunflower, it is especially suitable for dry and salty areas where other oilseeds are difficult to grow²⁴. In previous studies, Safflower appeared to be tolerant to heavy metal stress^{14,15}. In the present paper, we have conducted a study of the effect of excess FeSO₄ on the activities of some antioxidative enzymes, catalase and peroxidase, in roots and shoots of safflower (*Carthamus tinctorius* L.) seedlings.

MATERIALS AND METHODS

Plant material

Seeds of Safflower (*Carthamus tinctorius* L.) were collected from the agricultural fields of Gajanur Sakrebyle, Shimoga district, Karnataka, India. The seeds were surface-sterilized with 1% sodium hypochloride solution for 10 minutes and thoroughly rinsed with sterile distilled water for three times and used throughout the experiments.

Stress conditions

In order to determine the effect of iron on activity of antioxidant enzymes, catalase and peroxidase, surface sterilized Safflower seeds were soaked for 24 h in different concentration (0, 50, 100, 150 and 200 mg/l) of Fe²⁺ solutions on a rotary shaker at 150 rpm under room temperature. Seeds immersed only in distilled water served as control. To evaluate the effect of iron on catalase and peroxidase activities in shoot and root of safflower treated and control seeds were sown in plastic pots each filled with sterilized coirpith (coconut compost). Each pot sown with 3 seeds is incubated at greenhouse conditions and watered regularly. Seedlings raised from control and treated seeds were carefully uprooted without causing any damage to root tissues at different time intervals, 7, 14, 21 and 28 days after sowing. Fresh roots and shoot parts were washed in running tap water, blot dried and used to evaluate the enzyme activity.

Enzyme extraction

The plant roots and shoots (3:1, buffer volume: fresh weight) were homogenized in a pre-chilled pestle and mortar with 100 mM cold potassium phosphate buffer (pH 6.8). The homogenate was centrifuged at 12,000×g for 10 min at 4°C. The supernatant was transferred to a micro tube for further analysis.

Protein Estimation

The protein contents of the extracts were determined by using the standard procedure of Bradford (Bradford, 1976) with Bovine Serum Albumin (BSA) (sigma, USA) as standard.

Enzyme assay

Catalase activity was determined from safflower seedlings using the method of Beers and sizer (1951) with minor modification. The 2.4 ml reaction mixture contained 100 mM K-phosphate buffer (pH 6.8) and 30% (w/v) H_2O_2 and the reaction was initiated by adding 100 µl of extracted enzyme. The decrease in absorbance at 240 nm was measured for 3 min. The specific activity of catalase was expressed as change in optical density (OD) at 240 nm/min/mg protein. Experiments were conducted in triplicates and repeated in four weeks.

The substrate for peroxidase (POX) assay was prepared by adding 125 μ l of guaiacol and 153 μ l of H₂O₂ (30%) in 50 ml 100 mM potassium phosphate buffer (pH 6.8). The reaction was initiated by adding 100 μ l of crude enzyme extract to 2.4 ml of substrate buffer. Increase in absorbance was recorded spectrophotometrically for 3 min at 470 nm. Peroxidase specific activity was calculated as change in absorbance units/min/mg protein. Experiments were conducted in triplicates and repeated in four weeks.

Devi Prasad, A. G. et al Int. J. Pure App. Biosci. 2 (5): 118-123 (2014) ISSN: 2	2320 - 7051
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Statistical analysis

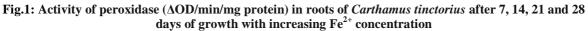
All data from laboratory and greenhouse experiments were analyzed separately for each experiment and were subjected to analysis of variance (ANOVA) (MINITAB, version 14). Significant effects of treatments were determined by the F values ($P \le 0.01$). Treatment means were separated using Tukey's HSD test.

RESULTS

Effects of iron on peroxidase activity

The activity of peroxidase showed a significant decrease ($p \le 0.01$) from 7th to 14th day of growth with increase in Fe²⁺ treatment levels (Fig.1) and then was maintained at relatively constant values from 14th day to 28th day of growth. Roots showed higher level of enzyme activity than shoots in Fe- treated as well as control plants.

peroxidase activity increased in shoots at all levels of Fe^{2+} treatments from 7th to 28th day. A 200 mg.l¹⁻ Fe²⁺ treatment led to about 1.3- 3.0 fold increase in peroxidase activity in shoots of safflower seedlings at 28th day of growth (Fig.2).



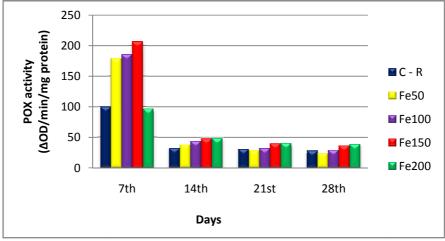
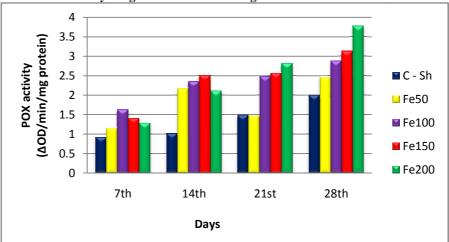


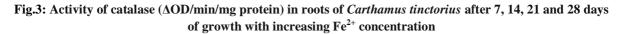
Fig.2: Activity of peroxidase (ΔOD/min/mg protein) in shoots of *Carthamus tinctorius* after 7, 14, 21 and 28 days of growth with increasing Fe²⁺ concentration



Devi Prasad, A. G. et al Int. J. Pure App. Biosci. 2 (5): 118-123 (2014)

Effects of iron on CAT activity

In Fe- treated seedlings a decrease in catalase activity was observed in roots during 28 days period (Fig.3). In shoots, up to 21^{st} day, the activity of CAT gradually decreased whereas at 28^{th} day a significant (p ≤ 0.01) increase in enzyme activity was noticed (Fig.4).



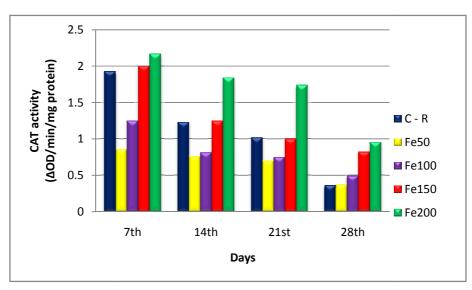
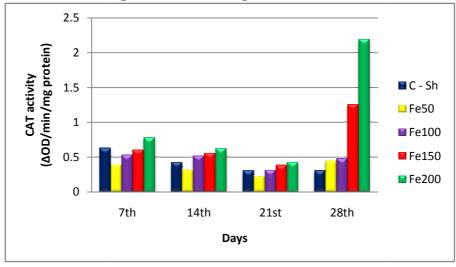


Fig.4: Activity of catalase (ΔOD/min/mg protein) in shoots of *Carthamus tinctorius* after 7, 14, 21 and 28 days of growth with increasing Fe²⁺ concentration



DISCUSSION

Plants growing in heavy metal-contaminated condition need to develop some degree of tolerance to metal toxicity for surviving. Induction in the activities of antioxidative enzymes is a general strategy for plants to overcome oxidative stress²⁵. The antioxidant enzymes and certain metabolites present in plants play an important role leading to adaptation and ultimate survival of plants during periods of stress^{8,16}. The present study suggested that Fe toxicity induces some of the key enzymes of antioxidant defense system such as catalase and peroxidase in roots and shoots of C. tinctorius.

POX belongs to enzymes involved in growth, development and senescence processes of plants. They affect lignin and ethylene synthesis, decomposition of indole-3-acetic acid (IAA), and involves in resistance against pathogens and wound healing².

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Devi Prasad, A. G. et al Int. J. Pure App. Biosci. 2 (5): 118-123 (2014) ISSN: 2320 – 7051

Catalase (CAT) belongs to most important enzymes scavenging the active oxygen species in plant cells. CAT participates in the main defense system against accumulation and toxicity of hydrogen peroxide and can play the role in controlling H_2O_2 level in cells. It acts on H_2O_2 and converts it to water and oxygen¹⁹. The activity of POX and CAT of *C. tinctorius* treated with Fe²⁺ mainly displayed biphasic responses with increased metal concentration. When safflower seeds were faced with low-level metal stress, it could activate POX and CAT activities, which lead to strengthening of seedlings to scavenge ROS while the activities decreased distinctly under too acute stress which overloaded cellular defense system of plants. Also the decline in the activities of POX and CAT might be due to the formation of protein complex with metals, to change the structure or integrity of proteins^{7,17}. For seeds treated with 200 mg.l⁻¹ of Fe²⁺, CAT and POX activities appeared to be inhibited.

Increased Fe^{2+} concentrations intensified development of iron toxicity symptoms on roots and shoots. It has been demonstrated that excess Fe^{2+} treatment induces oxidative stress in plant tissues^{11,20,23}. In the present investigation, the excess of Fe^{2+} (100- 200 mg.l¹⁻) induced toxicity in root and shoot of *Carthamus tinctorius*. These results support the possibility of $FeSO_4$ -induced toxicity mediated through oxidative stress. The results indicate an enhancement in the activity of peroxidase in shoots, suggesting that this enzyme serves as an intrinsic defense tool to resist Fe-induced oxidative damage in safflower plants. Peroxidases are widely accepted as 'stress enzymes'⁹. As guaiacol peroxidases are located in cytosol, cell wall, vacuole and in extracellular spaces, increased peroxidase activity in Fe^{2+} stressed seedlings might be possibly due to increased release of peroxidases localized in the cell walls. Under sub-lethal salinity and metal toxicity conditions, level of peroxidase activity has been used as potential biomarker to evaluate the intensity of stress^{20,21}. The results show significant decreased POX activity in Fe-treated roots of safflower seedlings from 7th to 14th day, revealing efficient break down of peroxides up to 14th day of growth in safflower seedlings. The POX activity remained almost unchanged from 14th to 28th which support the possibility of efficient detoxification.

In the present work, CAT activity was inhibited in roots from 7th day to 28th day of growth under Fe²⁺ toxicity especially at higher concentrations, while it decreased gradually in shoots from 7th to 21st day and then increased at 28th day. The decline in CAT activity in Fe-treated safflower plants might be due to inhibition of enzyme synthesis or due to a change in the assembly of enzyme subunits under such conditions. Wei-Ching *et al.*,²³ reported no change in CAT activity under iron toxicity in rice leaves, while in some other plants a decline (sunflower) or enhancement (*Nicotina plumbagnifolia*) of CAT activity has been found^{11,22}. Thus, in the present experiments, CAT did not appear to be an efficient scavenger of H₂O₂ than POX. These inconsistent results regarding CAT activity might be due to differences in the plant organs studied, the durations, concentrations of metals utilized and the plant species under investigation.

In conclusion, our results indicated that the exposure of *C. tinctorius* to iron provoke pronounced responses of antioxidative systems which protects the plants to some extent against oxidative damage, but the direction of response was dependent on the intensity of the stress.

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Devi Prasad, A. G. et al Int. J. Pure App. Biosci. 2 (5): 118-123 (2014)

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